eg60411 <mark>Bio-</mark> I	Material Science
	Toshiharu Enomae
Professor, PhD, Pa	aper Device and Eco-friendly materials
20	G103, 10:10-11:25, Tuesday

B	Biomaterial Science (Schedule)			
#	Date	Content		
1	4/15	History of papermaking		
2	4/22	Pulps – Beating and fiber properties		
3	5/9, Fri	Pulps – Additives and functions		
4	5/13	Papermaking processes & interfiber bonding		
5	5/20	Paper- Structural and absorption properties		
6	5/27	Paper- Mechanical and optical properties		
7	6/3	Polysaccharide chemistry by Assoc Prof Akiko Nakagawa		
8-9	6/10, 17	Pulping science and technology by Professor Hiroshi Ohi		
10	6/24	Recent trend of paper science and technology		













![](_page_1_Figure_3.jpeg)

![](_page_1_Figure_4.jpeg)

![](_page_1_Figure_5.jpeg)

![](_page_1_Figure_6.jpeg)

![](_page_2_Figure_1.jpeg)

## Stock preparation – additives

- Paper quality control
- Size (sizing agent) water repellency
- Filler brightness and opacity
- **Strength agent** dry or wet strength
- **Dye** optical brightening agent (OBA)
- Paper manufacturing control
- Retention aid ex. aluminum sulfate fines, fillers, and size retained more
- Aintiseptic (preservative)

	Acidic paper	Non-acidic paper
Size (sizing agent)	Rosin (Abietic acid)	Alkyl ketene dimer (AKD), Alkenyl succinic anhydride (ASA)
Retention aid	Alminium sulfate (alum), deteriorates paper	Cationic polymer such as Polyamine-amide epichlorohydrin (PAE)
Filler	<mark>Clay</mark> , Titan dioxide, Talk	Calcium carbonate, Titan dioxide
pH at papermaking	4.5~5.5	7.5~8.5 (7 or slightly greater)

![](_page_2_Figure_12.jpeg)

![](_page_2_Figure_13.jpeg)

![](_page_2_Figure_14.jpeg)

![](_page_3_Figure_1.jpeg)

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![](_page_4_Figure_1.jpeg)

![](_page_4_Picture_2.jpeg)

![](_page_4_Figure_3.jpeg)

![](_page_4_Figure_4.jpeg)

- → Ionic bond = a chemical bond in which two ions are joined together because one has a positive charge and the other a negative charge
- Covalent bond = a chemical bond in which two atoms share one or more pairs of electrons that hold them together (ca. 500 kJ/mol)
- Hydrogen bond = a weak connection that is formed between an atom of hydrogen (= a gas) and an atom of another substance such as oxygen or nitrogen (= a gas) (10~40 kJ/mol)
- Van der Waals forces = the relatively weak attractive forces that act on neutral atoms and molecules and that arise because of the electric polarization induced in each of the particles by the presence of other particles
- Dipolar bond (coordination bond)
- Metallic bond

![](_page_4_Figure_11.jpeg)

![](_page_4_Figure_12.jpeg)

![](_page_5_Figure_1.jpeg)

![](_page_5_Figure_2.jpeg)

![](_page_5_Figure_3.jpeg)

![](_page_5_Figure_4.jpeg)

- Auvalitages over internal sizing
- $\scriptstyle \bullet$  100% retained. No deposit on walls of former or froth

![](_page_5_Figure_7.jpeg)